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Jian-Chao Liu, He-Lian Chen, Ting Chen, Hong-Wu He* and Ming-Wu Ding

Department of Chemistry, Central China Normal University, Wuhan 430079, People's Republic of China

Correspondence e-mail: he1208@public.wh.hb.cn

Key indicators

Single-crystal X-ray study T = 292 K Mean σ (C–C) = 0.003 Å Disorder in main residue R factor = 0.052 wR factor = 0.156 Data-to-parameter ratio = 17.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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3-(4-Chlorophenyl)-5-methyl-2-propylamino-8,9,10,11-tetrahydro-2-benzothieno[2',3';2,3]pyrido[4,5-*d*]pyrimidin-4(3*H*)-one

The asymmetric unit of the title compound, $C_{23}H_{23}ClN_4OS$, contains two crystallographically independent molecules in which the orientations of the propylamine group and chlorophenyl rings with respect to the fused ring system are different. The terminal cyclohexene ring adopts a half-chair conformation. The crystal packing is stabilized by $N-H\cdots O$, $N-H\cdots N$ and $C-H\cdots O$ hydrogen bonds and $\pi-\pi$ interactions.

Comment

Many pyridopyrimidines exhibit pharmaceutical and germicidal activities (Anderson & Broom, 1977). Over two hundred tetrahydro- and octahydropyrido[4,3-*d*]pyrimidines have been synthesized as potential diuretic, antirheumatic and bacteriostatic drugs, but only a few fully aromatic pyrido[4,3-*d*]pyri-]pyrimidines are known. An important synthetic route for pyrido[4,3-*d*]pyrimidine is the condensation reaction of 4aminonicotinic acid and amines (Ismail & Wibberley, 1967). However, this method often requires long reaction times. Recently, we have developed a new and facile regioselective annulation process, which proceeded smoothly under mild conditions *via* a tandem aza-Wittig and cyclization reaction, to synthesize novel pyridopyrimidine derivatives (Zhou *et al.*, 2005). In this paper, we report the structure of the title compound, (I).



The asymmetric unit contains two crystallographically independent molecules in which the orientations of the propylamine group and chlorophenyl rings with respect to the fused ring system are different (Fig. 1). The bond lengths and angles in the thieno[2,3-*b*]pyridine ring system (Table 1) are in agreement with the corresponding values observed in a similar compound (Patel *et al.*, 2003). The C–S bond lengths are longer than the values observed in both free thiophene (1.714 Å; Bonham & Momany, 1963), measured using electron diffraction, and thieno[2,3-*c*]pyridine [1.728 (1) and Received 15 July 2005 Accepted 2 September 2005 Online 14 September 2005

organic papers

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Figure 1

The asymmetric unit of (I), showing 50% probability displacement ellipsoids. Both components of the disordered propylamine group are shown.

1.731 (1) Å; Nerenz et al., 1997]. The C7-S1-C14 angle is narrower than that observed in free thiophene [92.2 (2) $^{\circ}$]. The C5-N4-C7 angles are smaller than 120° (Ghosh & Simonsen, 1993). The C=N bond lengths (Table 1) are longer than the typical C=N distance of 1.28 Å (Zhao *et al.*, 2005). In both independent molecules, the central aromatic ring system consisting of thiophene, pyridine and pyrimidine rings is essentially planar, and the terminal cyclohexene ring adopts a half-chair conformation. The crystal packing is stabilized by N-H···O, N-H···N and C-H···O hydrogen bonds (Table 2). The C15'-C20' ring and the N4/C3-C5/C7-C8 ring of the inversion-related molecule at (1 - x, 1 - y, 1 - z) are stacked with a centroid-centroid distance of 3.754 (1) Å, indicating a weak π - π stacking interaction.

Experimental

6-Methyl-4-amino-5-ethoxycarbonyltetrahydrobenzo[4,5]thieno-[3,2;5,6]pyridine was prepared according to the literature procedure (Veronese et al., 1995) in 64% yield. Iminophosphorane was also synthesized according to a literature method (Wamhoff et al., 1993) in 93% yield. To a solution of iminophosphorane (0.525 g, 1 mmol) in anhydrous CH₂Cl₂ (10 ml) was added 4-chlorophenyl isocyanate (1.1 mmol) under N₂ atmosphere at room temperature. The reaction mixture was left unstirred for 5 h, and then the solvent was removed under reduced pressure and Et₂O/petroleum ether was added to precipitate triphenylphosphine oxide. Removal of the solvent gave carbodiimide, which was used directly without further purification. To the solution of the carbodiimide prepared above in CH₂Cl₂ (10 ml) was added *n*-propylamine (1.1 mmol). The reaction mixture was stirred for 30 min, and then the solvent was removed and anhydrous ethanol (10 ml) with several drops of CH₃CH₂ONa in CH₃CH₂OH was added. The mixture was stirred for 11 h at room temperature, the solution was condensed and the residue was recrystallized from CH3CH2OH to give the title compound. Single crystals were obtained by evaporation of a methanol solution (m.p. 501-503 K). Analysis calculated for C₂₃H₂₃ClN₄OS: C 62.93, H 5.28, N 12.76%; found: C 63.26, H 5.23, N 12.45%. ¹H NMR (CDCl₃, TMS, 400 MHz): δ 0.91 (t, J = 7.2 Hz, 3H, CH₃), 1.62 (q, J = 7.2 Hz, 2H, CH₂), 1.92 (s, 4H, 2CH₂), 2.87 (s, 2H, CH₂), 2.94 (s, 3H, CH₃ of pyridyl), 3.29 (s, 2H, CH₂), 3.44 (m, 2H, NCH₂), 4.41 (s, 1H, NH),

7.27–7.60 (m, 4H, Ar-H). MS (EI, m/z, %): 439 (M^+ +1 35.14), 438 (100), 398 (14.08), 396 (17.21), 395 (24.23), 110 (15.44).

Crystal data

C23H23CIN4OS	Z = 4
$M_r = 438.96$	$D_x = 1.344 \text{ Mg m}^{-3}$
Triclinic, P1	Mo $K\alpha$ radiation
a = 11.8884 (6) Å	Cell parameters from 3870
b = 13.3784 (7) Å	reflections
c = 15.7892 (8) Å	$\theta = 2.3-26.9^{\circ}$
$\alpha = 114.054 \ (1)^{\circ}$	$\mu = 0.30 \text{ mm}^{-1}$
$\beta = 98.926 \ (1)^{\circ}$	T = 292 (2) K
$\gamma = 101.335 \ (1)^{\circ}$	Block, colourless
$V = 2169.42 (19) \text{ Å}^3$	$0.40 \times 0.30 \times 0.30 \text{ mm}$

9849 independent reflections

 $R_{\rm int} = 0.040$

 $\theta_{\rm max} = 27.5^{\circ}$

 $h = -15 \rightarrow 15$

 $k = -17 \rightarrow 17$ $l = -20 \rightarrow 20$

6468 reflections with $I > 2\sigma(I)$

Data collection

Bruker SMART CCD area-detector diffractometer φ and ω scans Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $T_{\min} = 0.891, T_{\max} = 0.917$ 25134 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0778P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.053$	+ 0.1323P]
$wR(F^2) = 0.156$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.03	$(\Delta/\sigma)_{\rm max} = 0.001$
9849 reflections	$\Delta \rho_{\rm max} = 0.32 \text{ e } \text{\AA}^{-3}$
574 parameters	$\Delta \rho_{\rm min} = -0.21 \text{ e} \text{ \AA}^{-3}$
H-atom parameters constrained	

Table 1

Selected geometric parameters (Å, °).

S1-C7	1.737 (2)	S1′-C7′	1.734 (2)
S1-C14	1.738 (3)	S1'-C14'	1.735 (3)
O1-C2	1.209 (2)	O1′-C2′	1.220 (2)
N2-C1	1.308 (2)	N2' - C1'	1.307 (2)
N2-C4	1.367 (3)	N2' - C4'	1.369 (3)
N4-C5	1.334 (3)	N4' - C5'	1.327 (3)
N4-C7	1.349 (3)	N4′-C7′	1.343 (3)
C7-S1-C14	91.00 (11)	C7′-S1′-C14′	91.06 (10)
C1-N2-C4	118.07 (18)	C1'-N2'-C4'	117.71 (17)
C5-N4-C7	116.82 (18)	C5' - N4' - C7'	116.56 (18)

Table 2 Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N3' - H3'B \cdots N4^{i}$	0.90	2.30	3.072 (3)	144
N3-H3···O1′ ⁱⁱ	0.86	2.22	2.912 (2)	138
$C13' - H13C \cdot \cdot \cdot O1^{iii}$	0.97	2.38	3.285 (3)	154

Symmetry codes: (i) -x + 1, -y + 1, -z + 1; (ii) x, y, z - 1; (iii) -x + 1, -y, -z.

In one of the independent molecules, the propylamine group is disordered over two orientations with occupancies of 0.515 (8) and 0.485 (8). The disorder was modelled with C–C and C···C distances restrained to 1.54 (1) and 2.51 (1) Å, respectively. All the H atoms were placed in idealized positions and constrained to ride on their parent atoms, with N-H = 0.86 and 0.90 Å, C-H = 0.93-0.97 Å, and with $U_{iso}(H) = 1.5U_{eq}(C)$ for methyl H atoms and $1.2U_{eq}(C,N)$ for other H atoms.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine

structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.

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References

- Anderson, G. L. & Broom, A. D. (1977). J. Org. Chem. 42, 997-1000.
- Bonham, R. A. & Momany, F. A. (1963). J. Phys. Chem. 67, 2474-2477.
- Bruker (1998). SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.

- Ghosh, R. & Simonsen, S. H. (1993). Acta Cryst. C49, 1031-1032.
- Ismail, A. G. & Wibberley, D. G. (1967). J. Chem. Soc. C, pp. 2613-2616.
- Nerenz, H., Grahn, W. & Jones, P. G. (1997). Acta Cryst. C53, 787-789.
- Patel, U. H., Dave, C. G., Jotani, M. M. & Shah, H. C. (2003). Acta Cryst. C59, 030-032.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Veronese, A. C., Callegari, R. & Morelli, C. F. (1995). *Tetrahedron*, **51**, 12277– 12284.
- Wamhoff, H., Herrmann, S., Stolbern, S. & Nieger, M. (1993). *Tetrahedron*, **49**, 581–594.
- Zhao, J. F., Wang, C. G. & Ding, M. W. (2005). Chin. J. Struct. Chem. 24, 439–443.
- Zhou, H. B., Cui, Z. P. & Liu, J. C. (2005). J. Central China Normal Univ. 39, 343–346.